

Use of isotopes in metabolomics

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2-22-17

Synopsis

- **Natural abundance isotopes**
- **Tracing a metabolic pathway**
 - Labeling a precursor for qualitative analysis
 - 95% isotope/5% unlabeled and 5% unlabeled/95% isotope
- **Following individual carbon atoms**
- **Quantitative analysis of metabolic flux**
- **Post-extraction isotopic labeling**

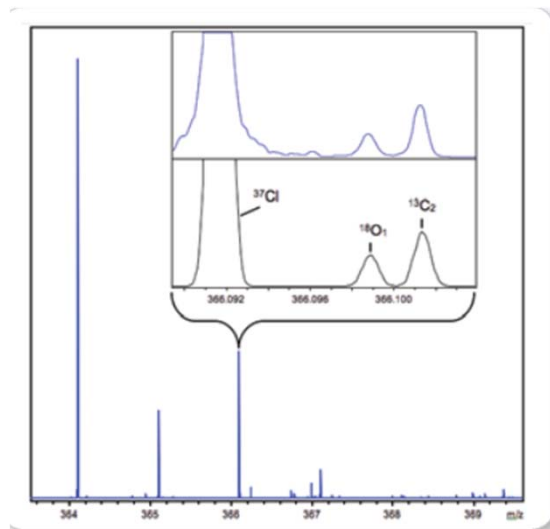
Value of natural isotopes

- The natural abundance of isotopes enables the investigator to determine the charge state of an ion
 - The principal contribution to $[M+H]^+$ or $[M-H]^-$ isotope ions comes from ^{13}C (~1.1% of all carbon atoms)
 - The intensity of the ^{13}C isotope ion increases relative to the number of carbon atoms
 - There is often an observable $^{13}\text{C}_2$ isotope peak

Value of the $[M+/-H+2]$ peak

- The mass difference due to a nominal increase in mass of 2 contains a lot of information
 - These are isotopic mass differences for each of the common elements
 - $^1\text{H}_2/{}^2\text{H}_2$ 2×1.006277 = 2.012554 (0.012%)
 - $^{12}\text{C}_2/{}^{13}\text{C}_2$ 2×1.003355 = 2.006710 (1.078%)
 - $^{14}\text{N}_2/{}^{15}\text{N}_2$ 2×0.997035 = 1.994079 (0.364%)
 - $^{16}\text{O}_2/{}^{17}\text{O}_2$ 2×1.004217 = 2.008434 (0.038%)
 - $^{16}\text{O}_2/{}^{18}\text{O}_1$ 1×2.004246 = 2.004246 (0.205%)
 - $^{32}\text{S}_2/{}^{33}\text{S}_2$ 2×0.999387 = 1.998774 (0.752%)
 - $^{32}\text{S}_2/{}^{34}\text{S}_1$ 1×1.995796 = 1.995796 (4.252%)

The importance of the M+2 ion

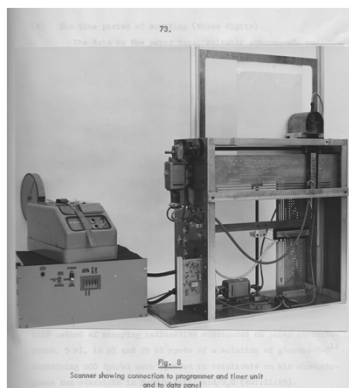


From Bruker

Using isotopes to trace a pathway

- Early studies (1930s) used ^2H , ^{13}C and ^{15}N labeling to map pathways
 - Limited to 1-200 m/z mass range
- 1950s/60s ^{14}C -radiotracers
 - 2D-Paper or thin layer chromatography
 - Radio gas chromatography
 - labeling of specific carbon atoms

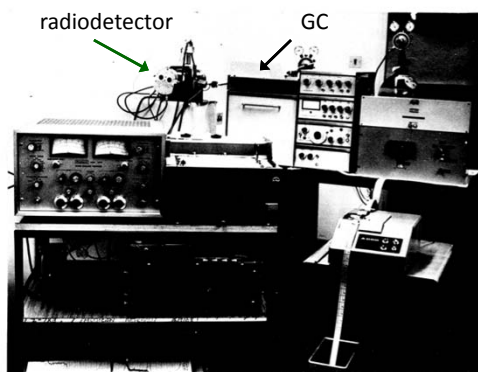
Origins of practical metabolomics Imperial College 1967-1970



Radio 2D-paper chromatography scanner with digitization of collected data

The room had 20 of these scanners – data analyzed by a central computer (in 1968)

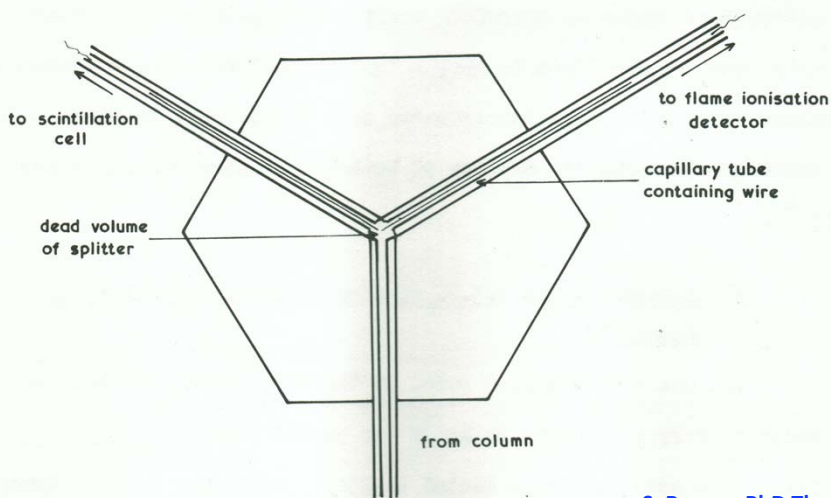
Courtesy of K.R. Mansford, PhD



Radio gas-liquid chromatography with digitization of collected data

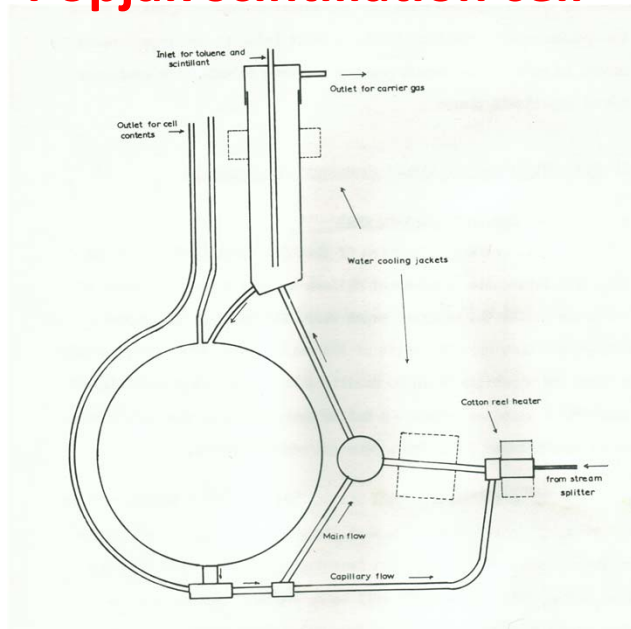
Developed this for my PhD work (1967-1970) to study glucose metabolism in acellular slime moulds

Stream splitter for radio GC

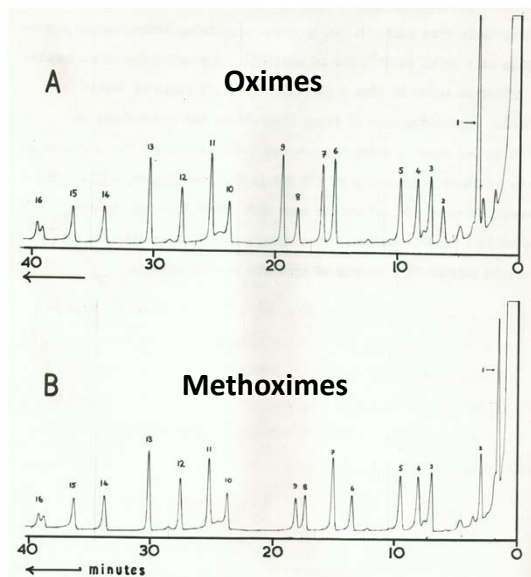


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Popjak scintillation cell



GC of glycolytic and Krebs cycle intermediates

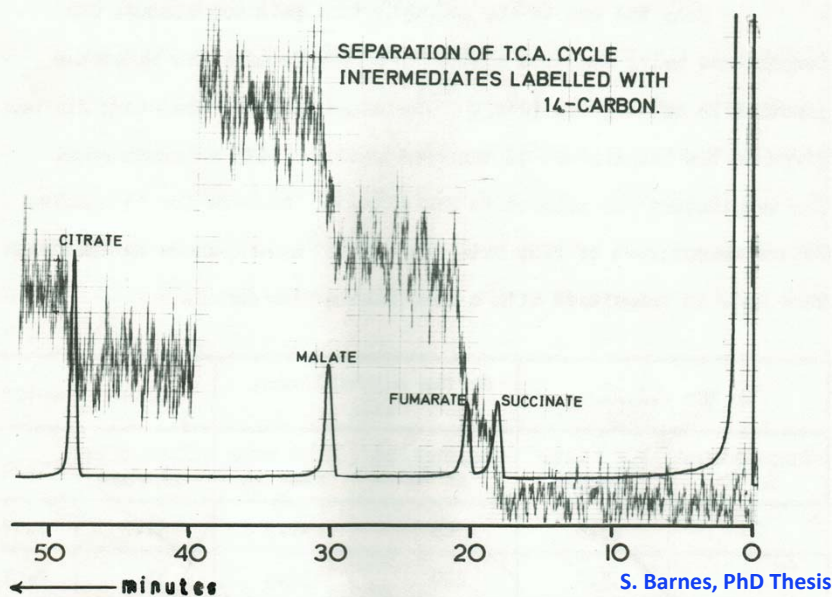


Temperature programming
of TMS ester/ethers on a 5' x
¼ inch packed column of
Chromosorb W coated with
OV-1 liquid phase

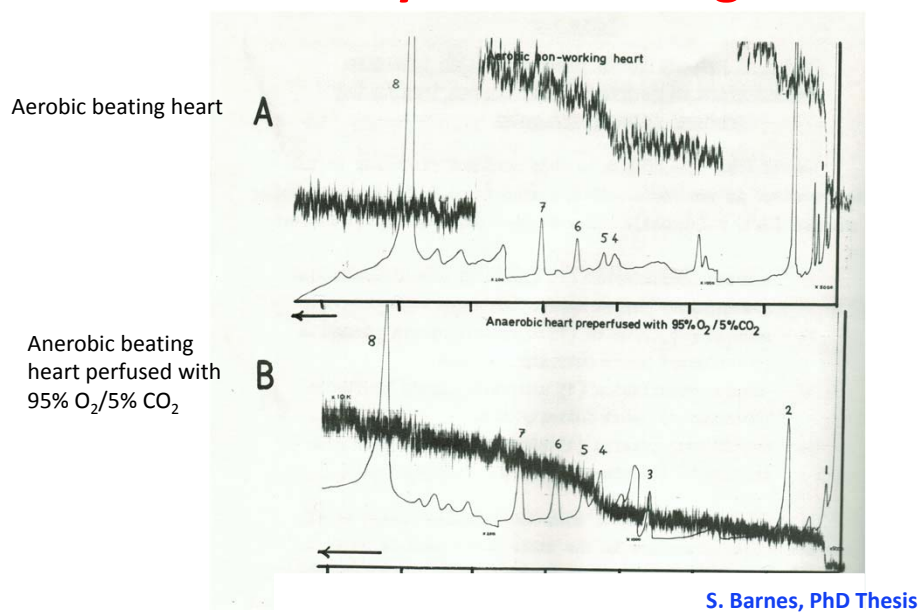
1=pyruvate, 2=??,
3=phosphate, 4=succinate,
5=fumarate, 6=oxaloacetate,
7=malate, 8=αKG,
9=hexadecane, 10=αGP,
11=citrate, 12=α-D-glucose,
13=β-D-glucose, 14=docosane,
15=F6P, 16=G6P

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Radio-GC of Krebs Cycle intermediates

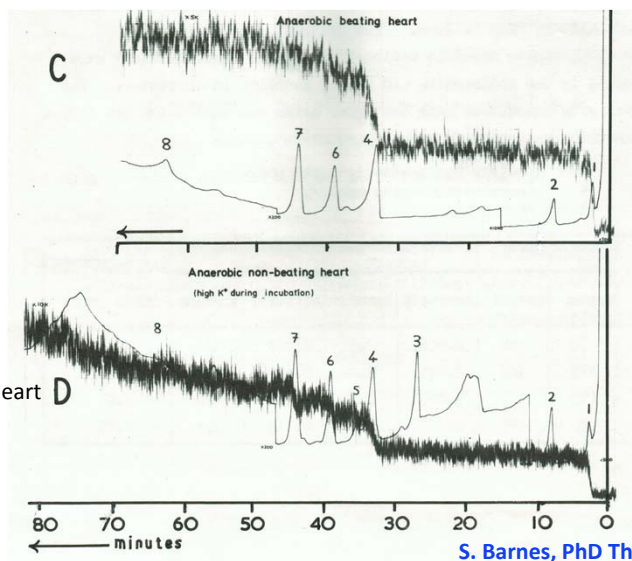


Radio GC analysis of beating heart



Radio GC analysis of anerobic heart

Anerobic beating heart

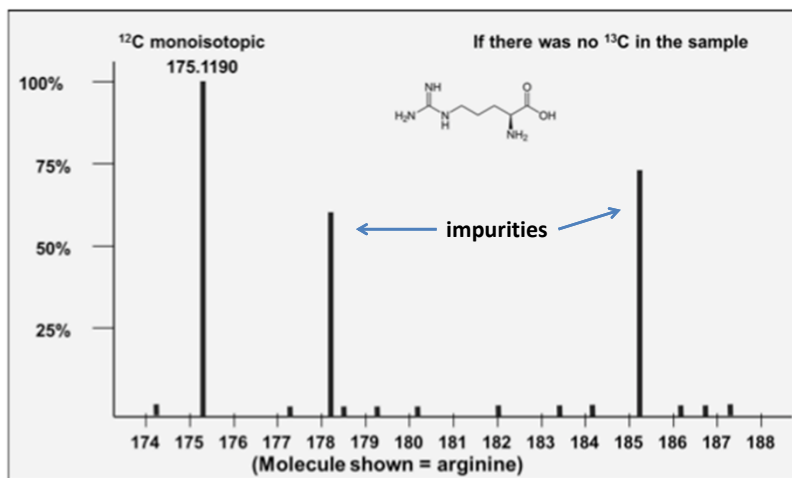


Anerobic non-beating heart

Tracking metabolites with IROA

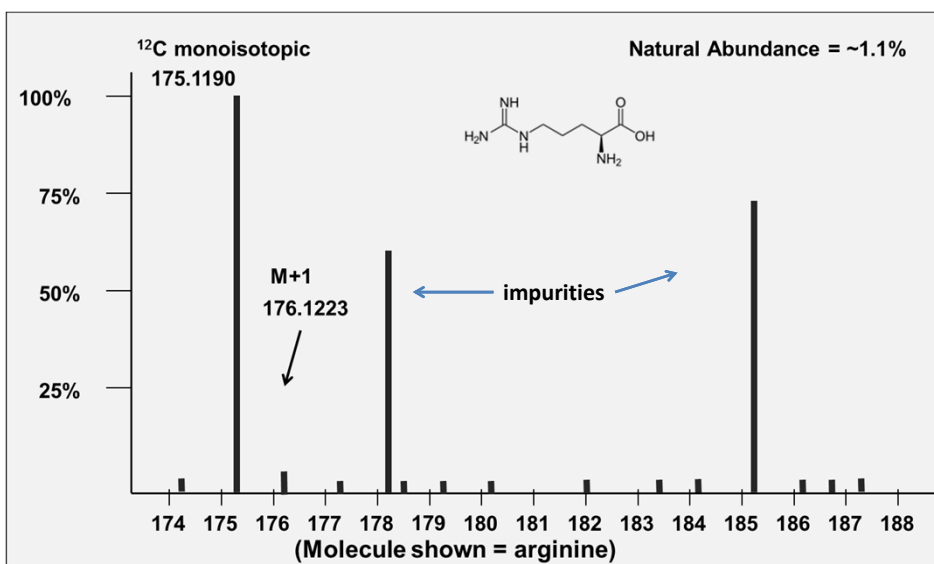
- **Isotope ratio outlier analysis (IROA)**
 - Not used for flux analysis, but rather to create a unique signal for metabolites
 - Used for LC-MS (and possibly GC-MS)
 - Designed to distinguish between metabolites of interest and background signals
 - Requires uniform labeling at the 95% and 5% ¹³C-enrichment levels

All ^{12}C in arginine $[\text{M}+\text{H}]^+$



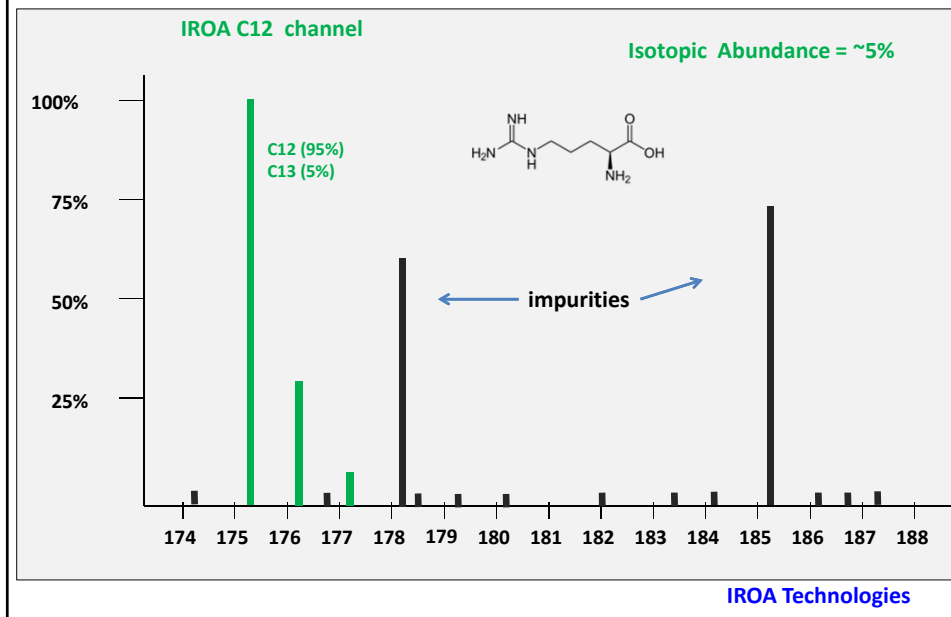
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Natural abundance of ^{13}C in arginine

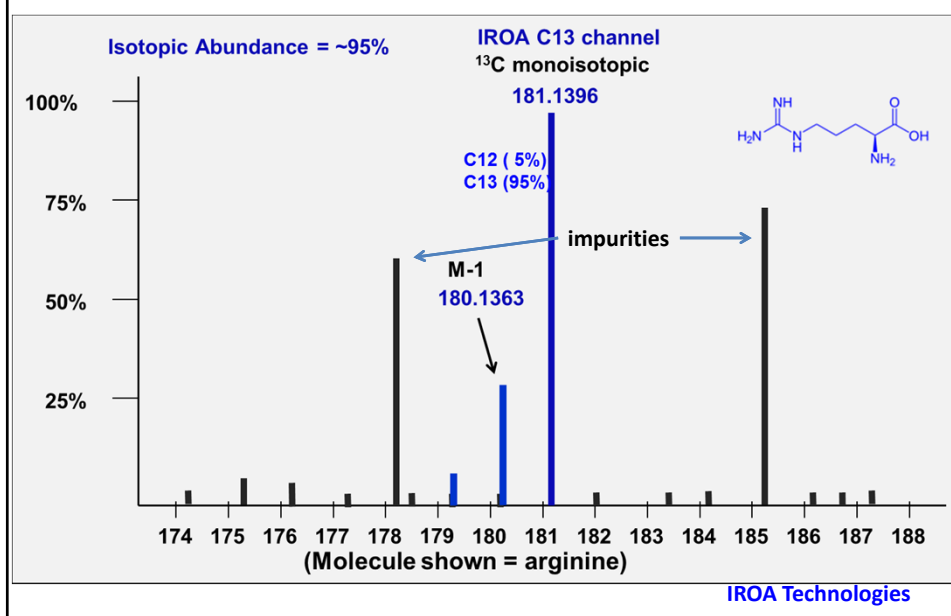


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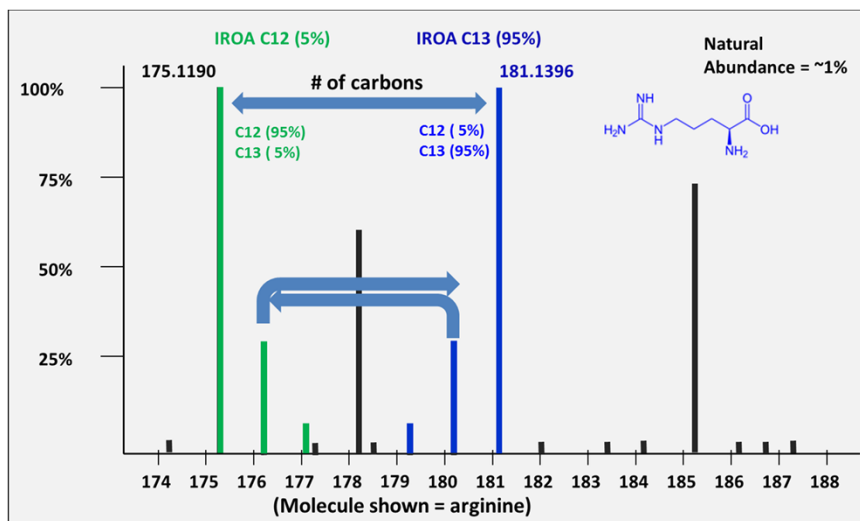
Making ^{13}C abundance = 5%



Making ^{13}C abundance = 95%

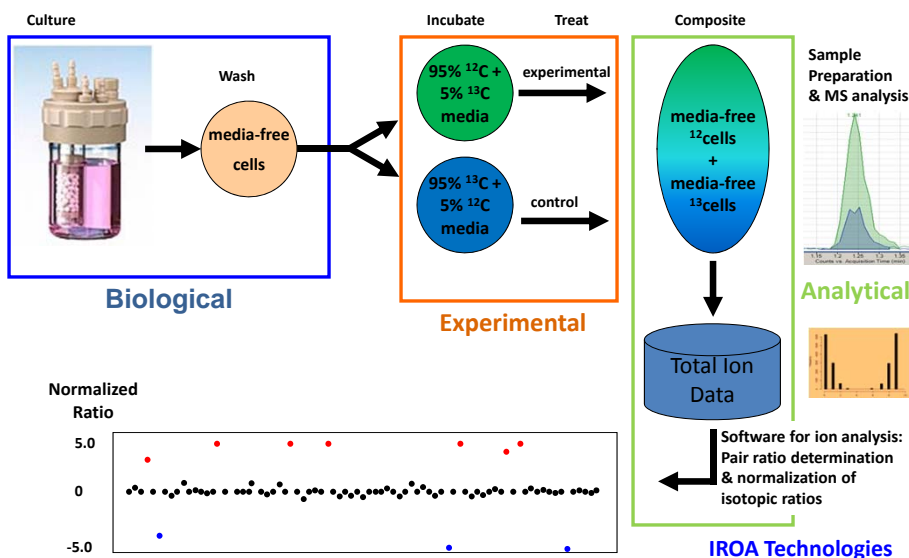


Pairing the 5% and 95% ^{13}C -labeling



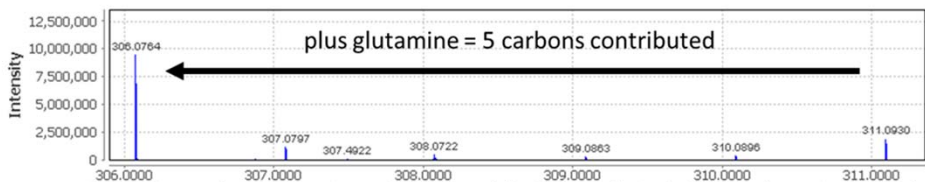
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The IROA approach

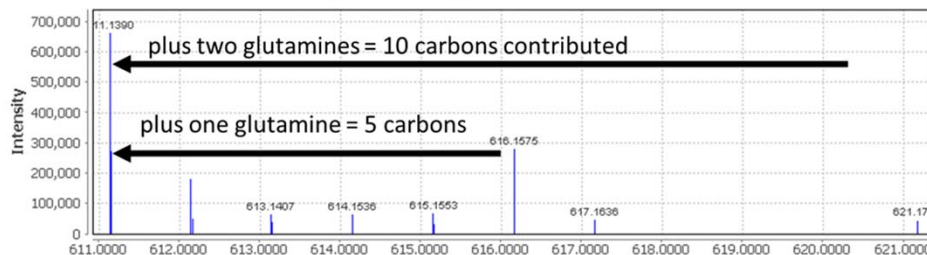


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The span of isotopes = # carbon atoms



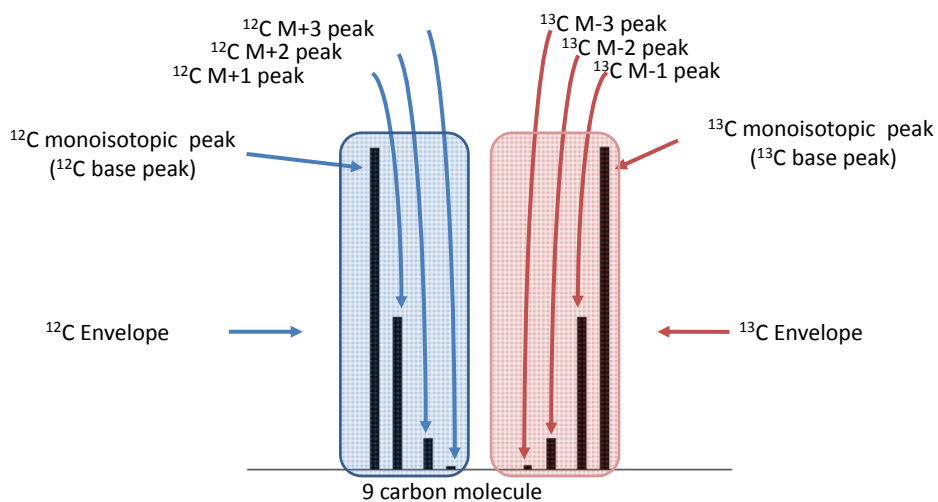
Glutathione shows the sizes of the non-labeled, and singly labeled species, which inherits a five carbon fragment from glutamine.



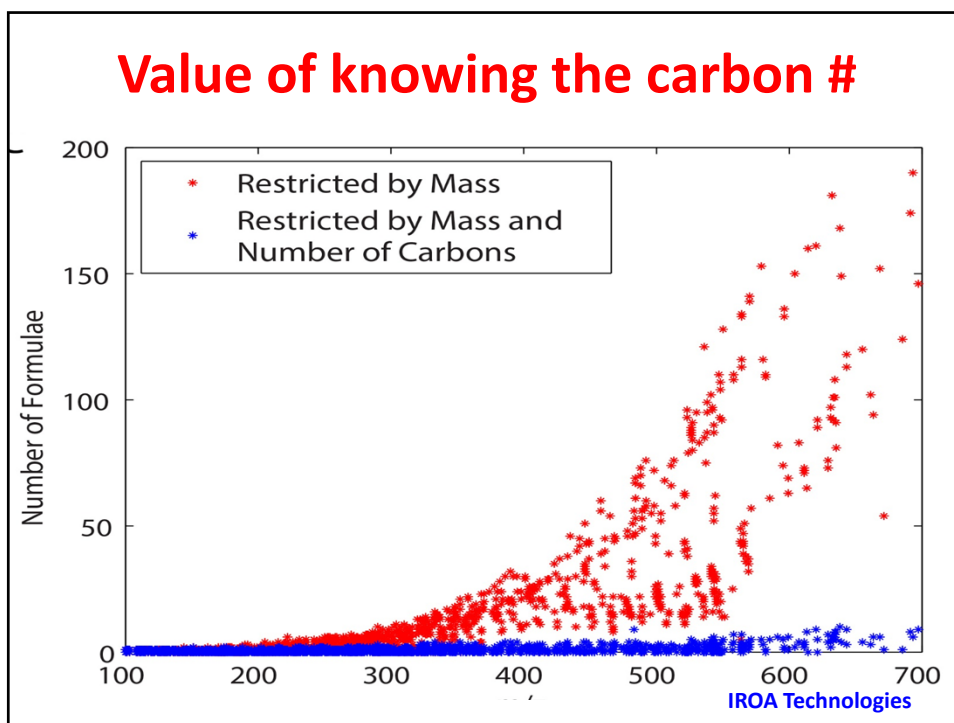
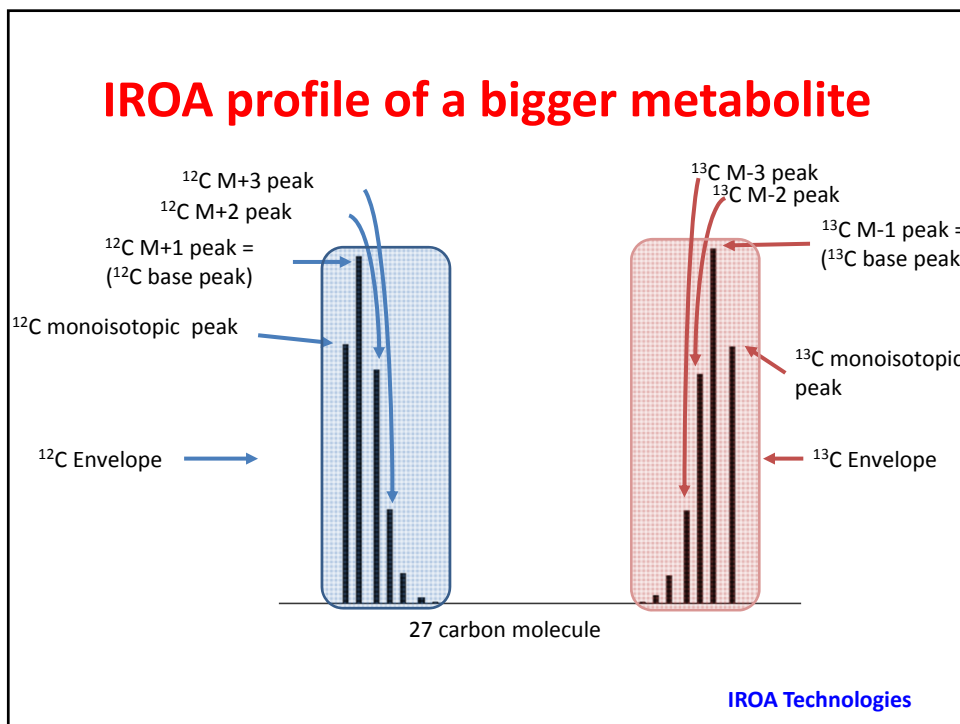
Oxidized glutathione shows the sizes of the non-labeled, singly labeled, and doubly labeled species. Each inheriting a five carbon fragment from glutamine.

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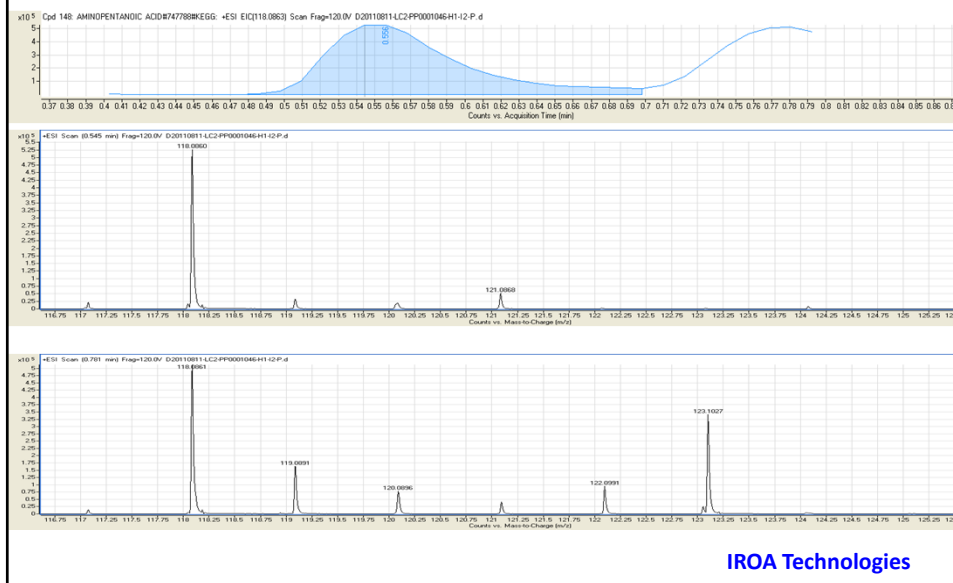
The IROA peak pattern



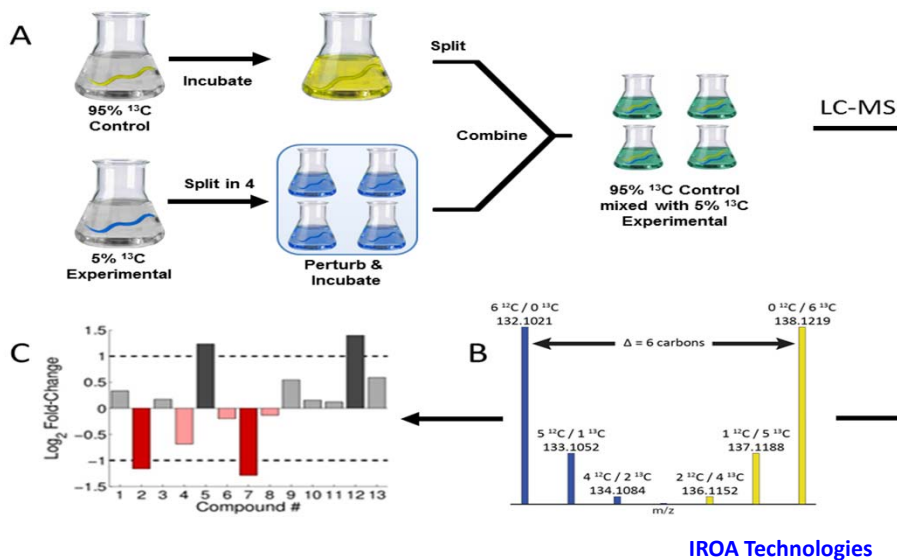
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Avoiding metabolite artifacts with IROA



IROA with *C. elegans*



Effect of a toxin on *C. elegans*

- **742 strong IROA peak pairs were found**
 - 314 named / 428 formula determined
 - Positive and negative mode LC
 - Thermo Orbi-trap @ 70K resolution
- **Strong response signature determined**
 - Basic statistics, PCA, Random Forest, NMF, SOM
 - 74 compounds were considered significant by at least 3 of these methods.

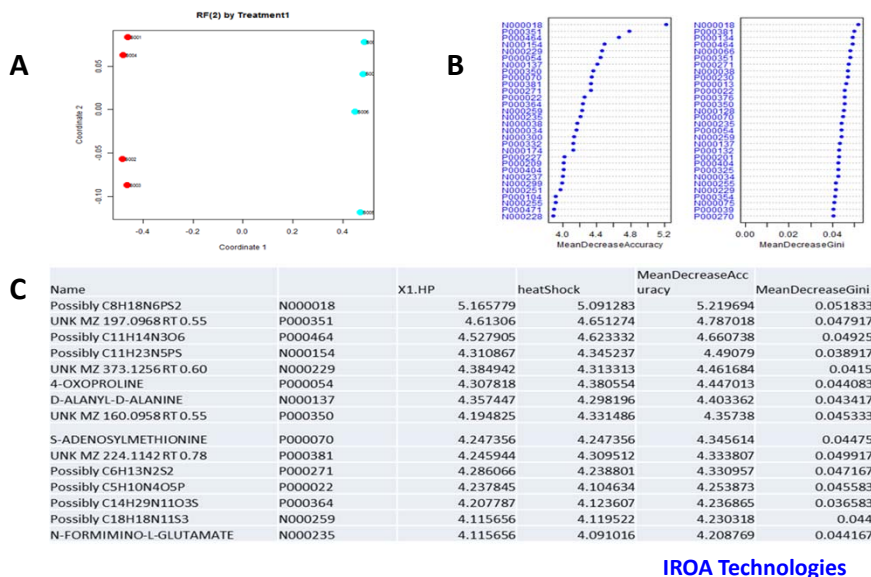
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Ions significantly affected by the toxin

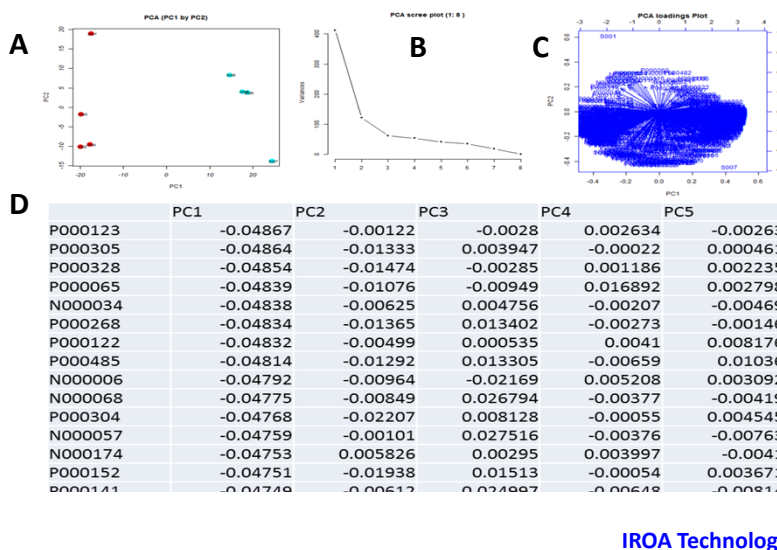
Name		p.value	F-Value treatment ²
L-LYSINE	P000009	7.89E-05	89.71
Possibly C ₂ H ₇ N ₃ O ₉ S	P000018	3.06E-05	124.99
L-ARGININE	P000019	0.000131	74.84
Possibly C ₂ H ₉ NO ₁₁	P000025	0.000182	66.63
UNK <i>m/z</i> 369.2215 RT 0.58	P000040	2.19E-05	140.24
SACCHAROPINE	P000046	7.23E-05	92.51
L-THREONINE	P000051	2.64E-05	131.52
L-GLUTAMIC ACID	P000053	1.09E-06	389.79
4-OXOPROLINE	P000054	1.74E-05	151.81
Possibly C ₂ H ₂ NO	P000058	1.8E-05	150.26
L-VALINE	P000060	0.000262	58.37
CITRULLINE	P000061	3.15E-05	123.67
4-METHYLENE-L-GLUTAMINE	P000062	0.000169	68.40
L-METHIONINE S-OXIDE	P000065	7.55E-06	202.32
L-PROLAMIDE	P000085	0.000227	61.56
STACHYDRINE	P000102	4.75E-05	107.19
UNK <i>m/z</i> 206.0368 RT 0.71	P000114	0.000251	59.35
N-ACETYLPUTRESCINE	P000122	8.96E-07	417.06
EPSILON-CAPROLACTAM	P000123	1.29E-08	1731.72
2-AMINO-OCTANOIC ACID	P000131	0.000213	62.99
UNK <i>m/z</i> 345.1258 RT 0.97	P000141	0.000111	79.36
Possibly C ₁₀ H ₁₉ N ₂ O ₅ P ₂	P000151	0.000154	70.78
CYS-GLY	P000152	0.000116	78.29
URATE	P000156	0.000222	62.02
Possibly C ₁₃ H ₁₆ N ₂ OPS	P000218	1.1E-05	177.82

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Multivariate statistics



PCA analysis of toxin's effect



Summary of most likely metabolites

Name		Stats1	RFTop1	RFTop2	NMF3	NMF4	NMF5	NMF6	Count
UNK <i>m/z</i> 160.0958 RT 0.55	P000350	1	1	1	1	1	1	1	7
UNK <i>m/z</i> 197.0968 RT 0.55	P000351	1	1	1	1	1	1	1	7
UNK <i>m/z</i> 216.0852 RT 0.61	N000034	1	1	1	1	1	1	1	7
D-ALANYL-D-ALANINE	N000137	1	1	1	1	1	1	1	7
Possibly C ₂₅ H ₃₄ N ₄ O ₂	N000174	1	1	1	1	1	1	1	7
UNK <i>m/z</i> 373.1256 RT 0.60	N000229	1	1	1	1	1	1	1	7
2-AMINO-OCTANOIC ACID	P000131	1	1	0	1	1	1	1	6
Possibly C ₆ H ₈ N ₂ O ₃	P000354	1	1	0	1	1	1	1	6
UNK <i>m/z</i> 510.2122 RT 0.68	P000373	1	1	0	1	1	1	1	6
UNK <i>m/z</i> 224.1142 RT 0.78	P000381	0	1	1	1	1	1	1	6
Possibly C ₆ H ₈ NO ₃ P	P000410	1	1	0	1	1	1	1	6
Possibly C ₁₁ H ₁₄ N ₂ O ₆	P000464	0	1	1	1	1	1	1	6
Possibly C ₆ H ₈ N ₂ O ₃ P	P000471	1	1	0	1	1	1	1	6
Possibly C ₆ H ₁₂ N ₂ O ₃ PS	N000006	1	1	0	1	1	1	1	6
Possibly C ₁₁ H ₁₂ N ₂ PS	N000154	1	1	1	0	1	1	1	6
D-GLUCOSE	N000228	1	1	0	1	1	1	1	6
UNK <i>m/z</i> 548.2037 RT 0.63	N000232	1	1	0	1	1	1	1	6
GLYCERATE	N000237	1	1	0	1	1	1	1	6

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